

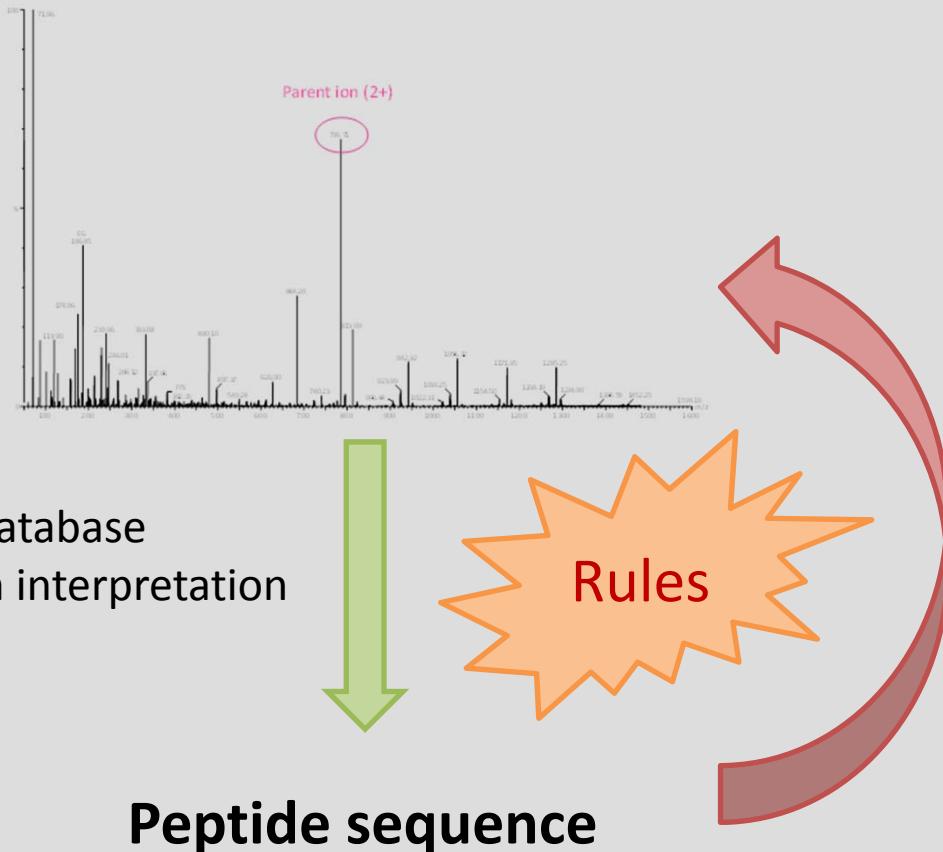
Peptide fragmentation : sequence and size effects on peptide fragmentation

Guillaume van der Rest¹, Renjie Hui², Gilles Frison²,
Julia Chamot-Rooke²

¹*Laboratoire de Chimie Physique, Université Paris-Sud*

²*Laboratoire des Mécanismes Réactionnels, Ecole
Polytechnique*

Why fragmentation rules are essential

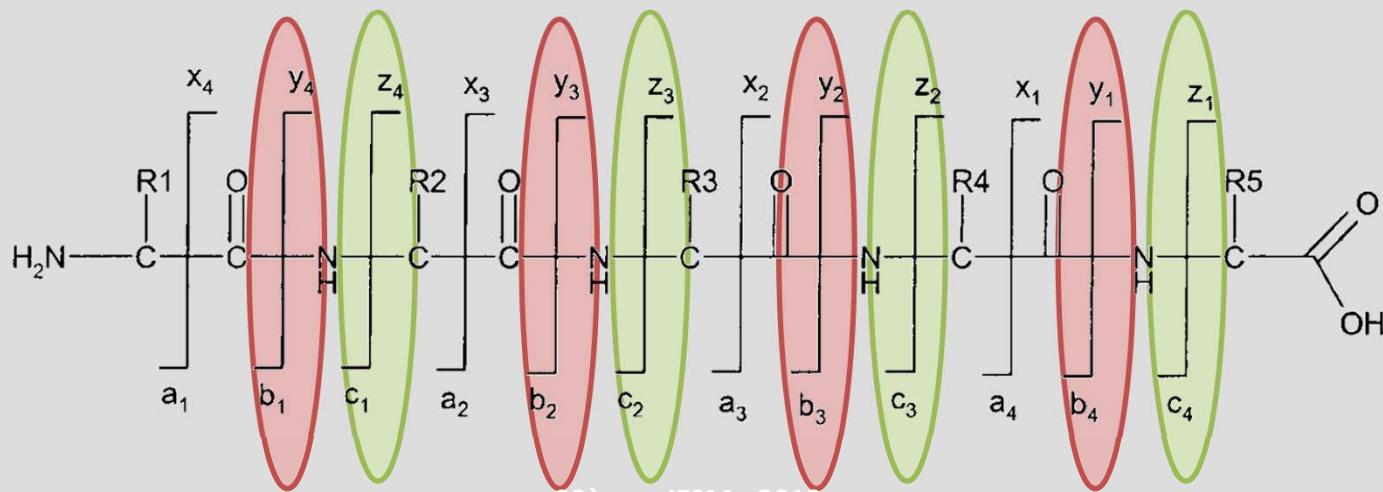


How efficient are we at:

- Predicting peak m/z ?
- Predicting intensities?

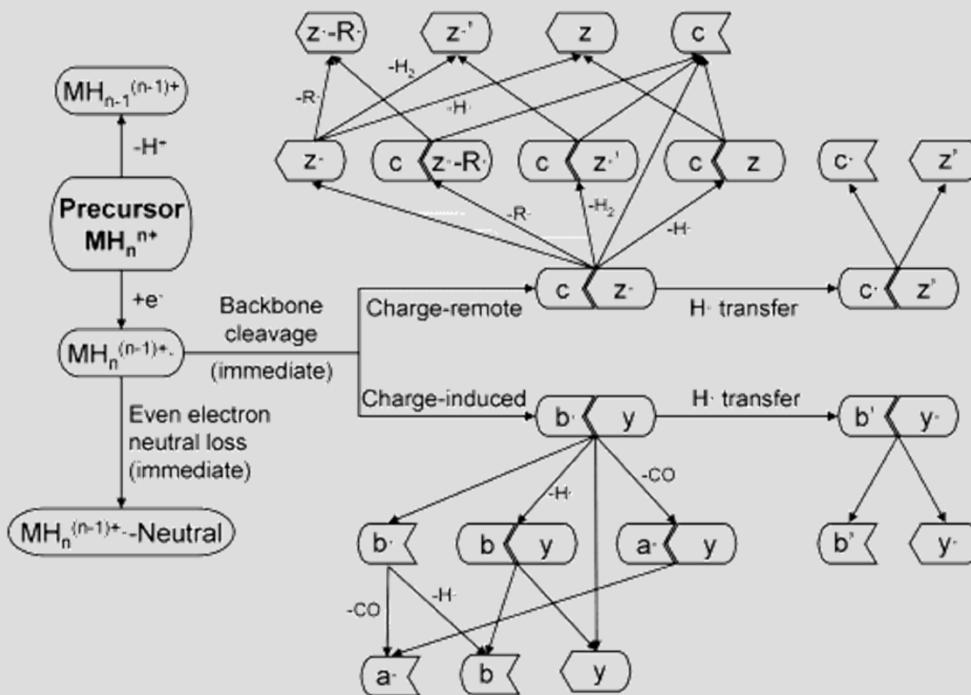
Rules can be simple...

- In CID :
 - Fragmentation of peptides lead to *b* and *y* type ions
- In ECD :
 - Fragmentation of peptides lead to *c* and *z* type ions



Or include a high number of parameters

- Reflect the diversity of protein and peptide compositions



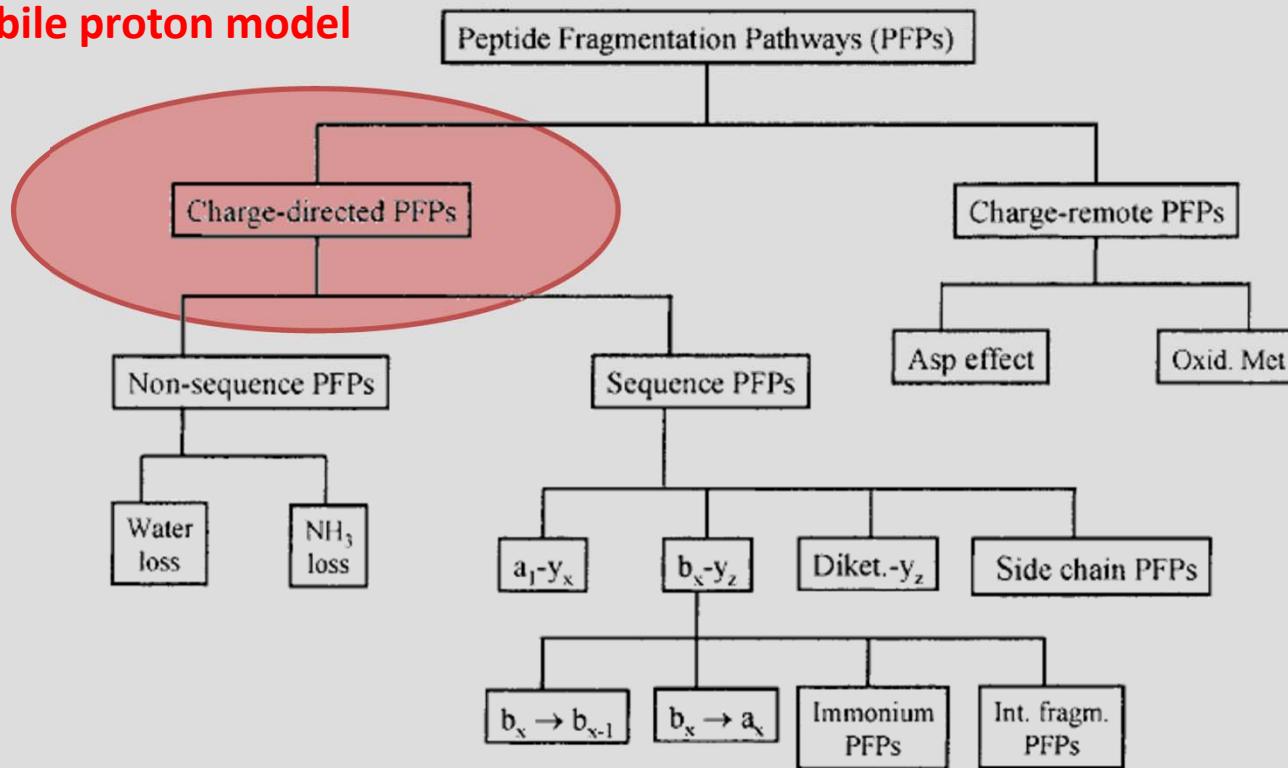
280 parameters, including

- Gas-phase basicity of residue n
- Hydrogen affinity of residue n
- Same for residue n+1

Z. Zhang, *Anal. Chem.* 1990 (2010)

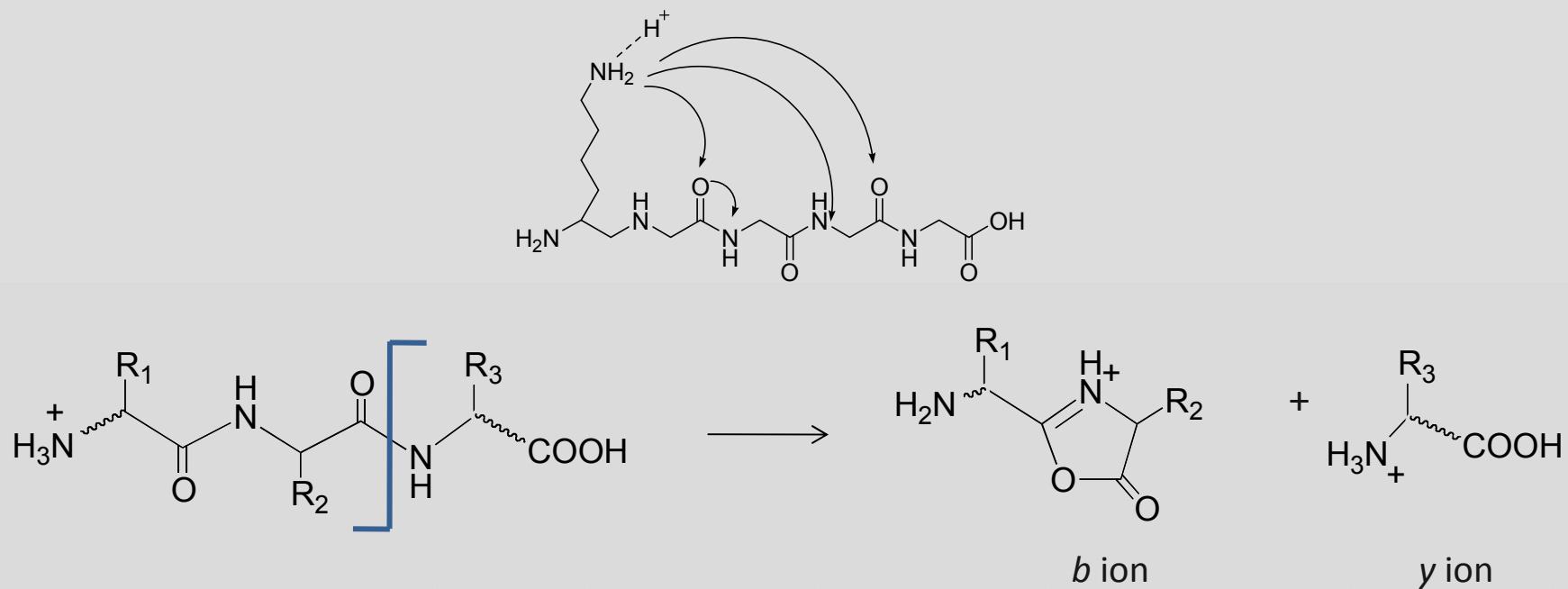
Low-energy CID of peptides

Mobile proton model



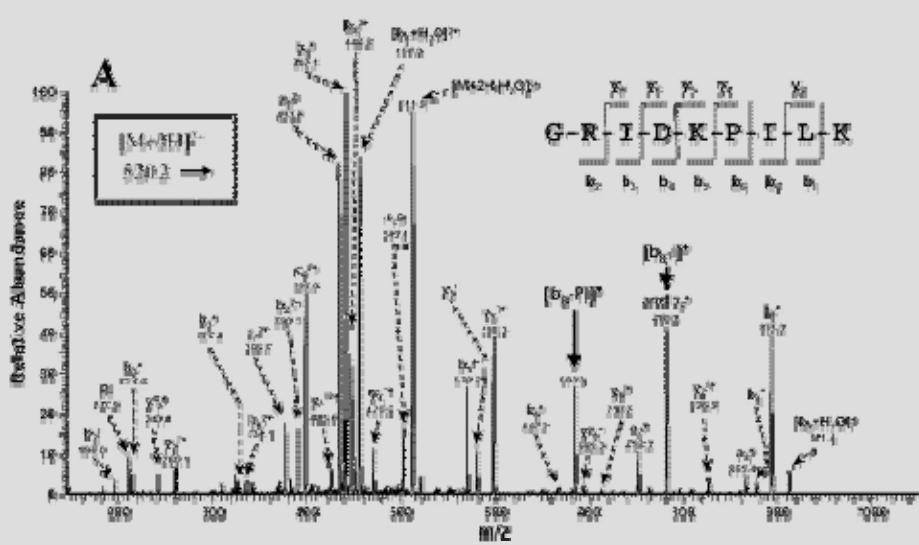
Paisz, Suhai, *Mass Spectrom. Rev.* 508 (2005)

The mobile proton model

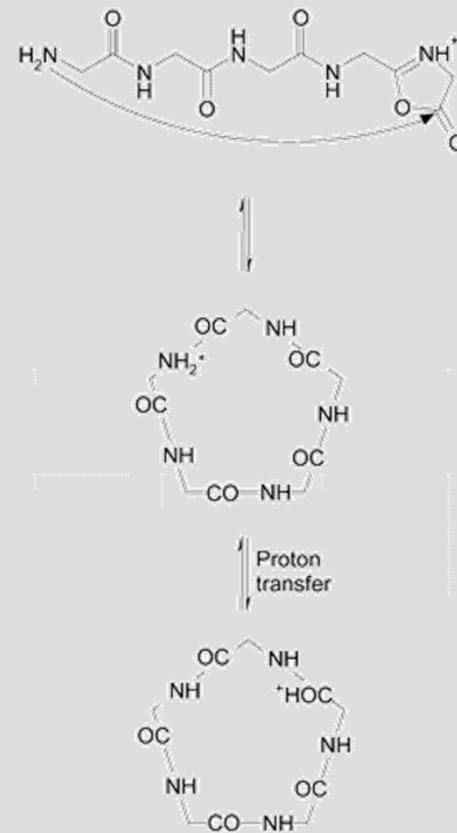


Dongré, *J. Am. Chem. Soc.* (1996)
Tsaprailis et al., *J. Am. Chem. Soc.* (1999)

Small intensity ions: a large number of origins



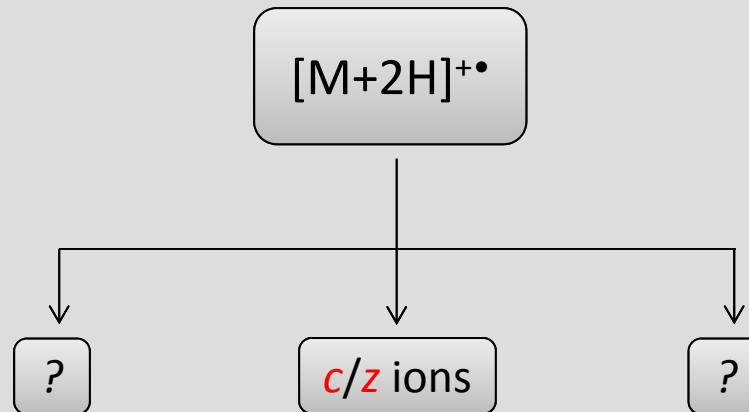
J. Yague *et al.* *Anal. Chem.*, 1524 (2003)



U. Erlekam, *J. Am. Chem. Soc.*, 11503 (2008)

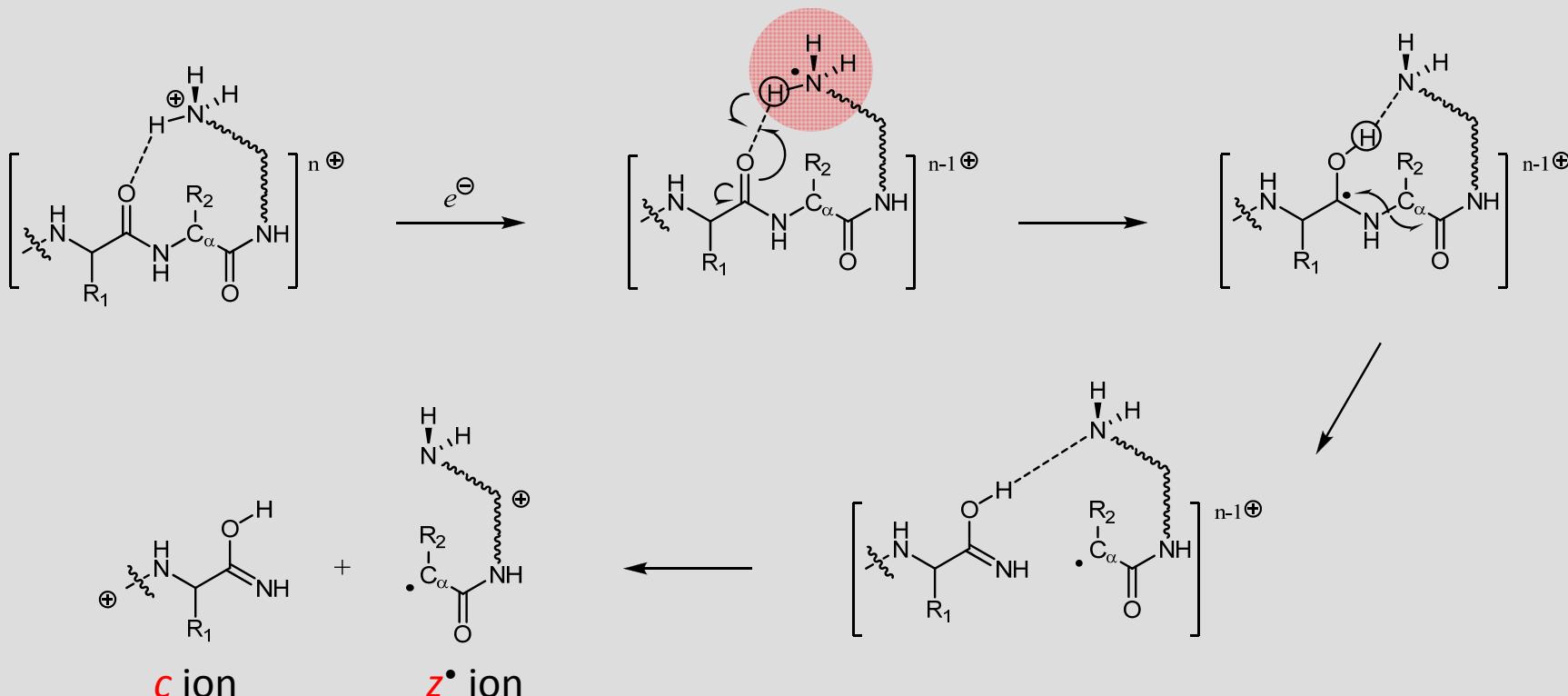
On the mechanism of ECD

- Other pathways than *c/z* fragmentation are observed.
 - Formation of *b*, *y*, *a*, *w* ions has been described.
- Little is known of the parameters favoring a given pathway.



Positive charge directed fragmentations (Cornell-like)

Electron capture at ammonium followed by H[•] transfer to a carbonyl.
Variants depend on structural changes before H[•] transfer.

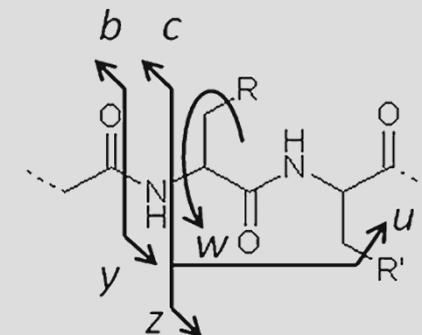
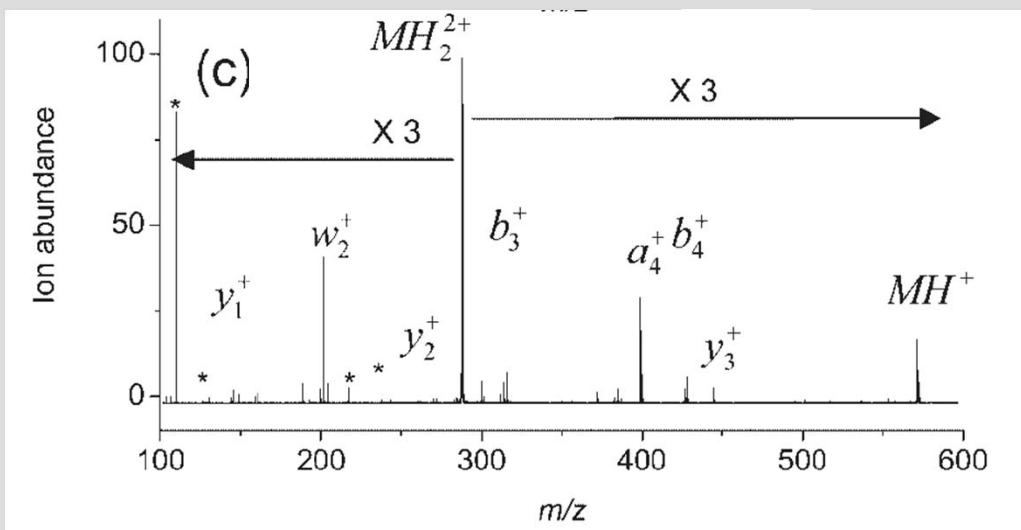


After covalent bond breaking

- H^\bullet transfers
 - within the z^\bullet fragment
 - Secondary fragmentation (side-chain loss, amine loss, ...)
 - Between c and z^\bullet fragments
 - Formation of $[c+\text{H}]^\bullet$ or $[z-\text{H}]$ depending on the intermediate $[c, z^\bullet]$ lifetime.
- Separation of the c / z moieties
 - Not systematic, can lead to a stable $[\text{M}, n\text{H}]^{(n-1)\bullet+}$ reduced species.

Three independent pathways in ECD fragmentation

- ECD de $[AGWLK+2H]^{2+}$



R. Antoine et al. RCMS 2006

- Pas de fragments c/z !
- Essentiellement des fragments w , b et y

b ions in ECD

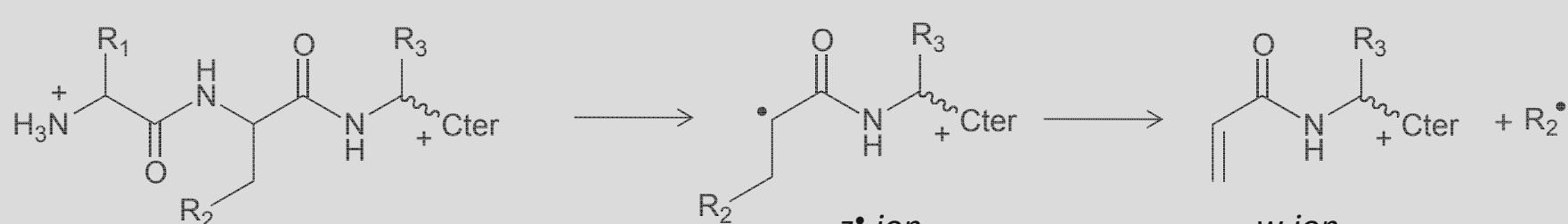
- *b* are frequently present in ECD spectra

- Possible origins for the formation of *b* ions:
 - Vibrational excitation of the $[M+nH]^{n+}$ precursors
 - Secondary fragmentation of *c* ions
 - H^\bullet loss after electron capture leading to a $[M+(n-1)H]^{(n-1)+}$ ion leading to *b/y* ions.

Liu and Hakansson. JASMS 2007

w ions

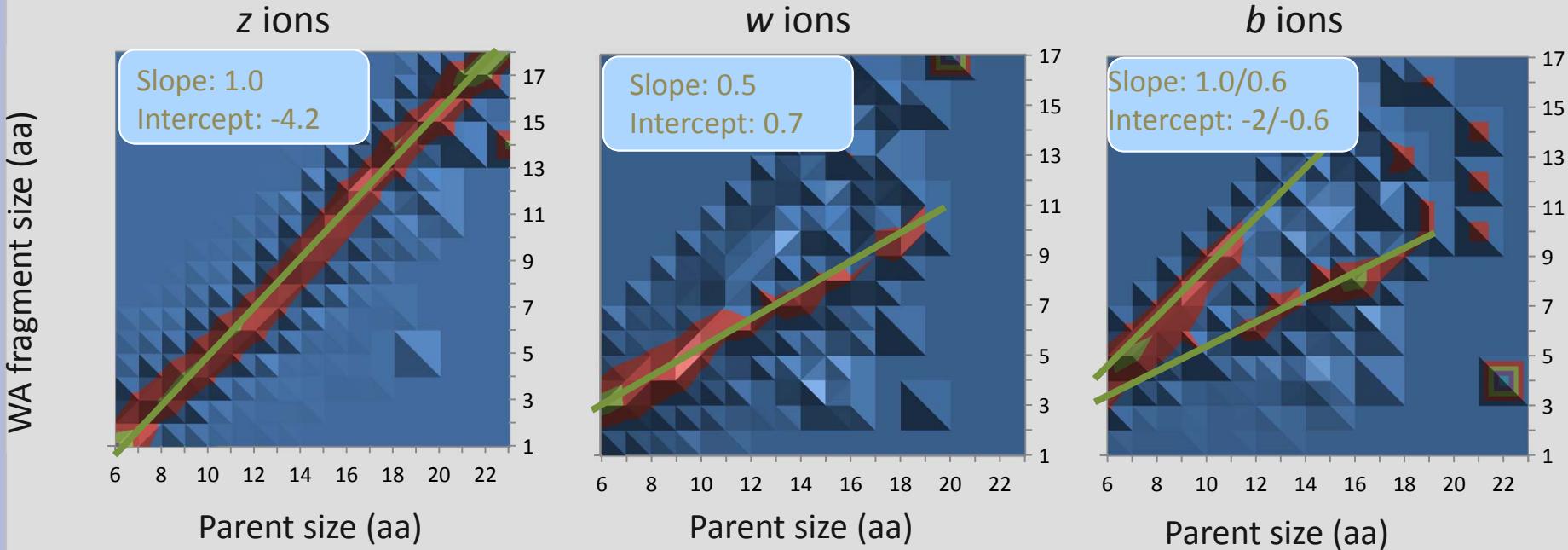
- Generally, w ions are considered secondary fragments of z ions
 - But no z ions are present in our ECD spectra.



R. Zubarev, *J. Am. Chem. Soc.* (2003)
R. Zubarev, *Anal. Chem.* (2003)
M. Savitski, *Anal. Chem.* (2007)

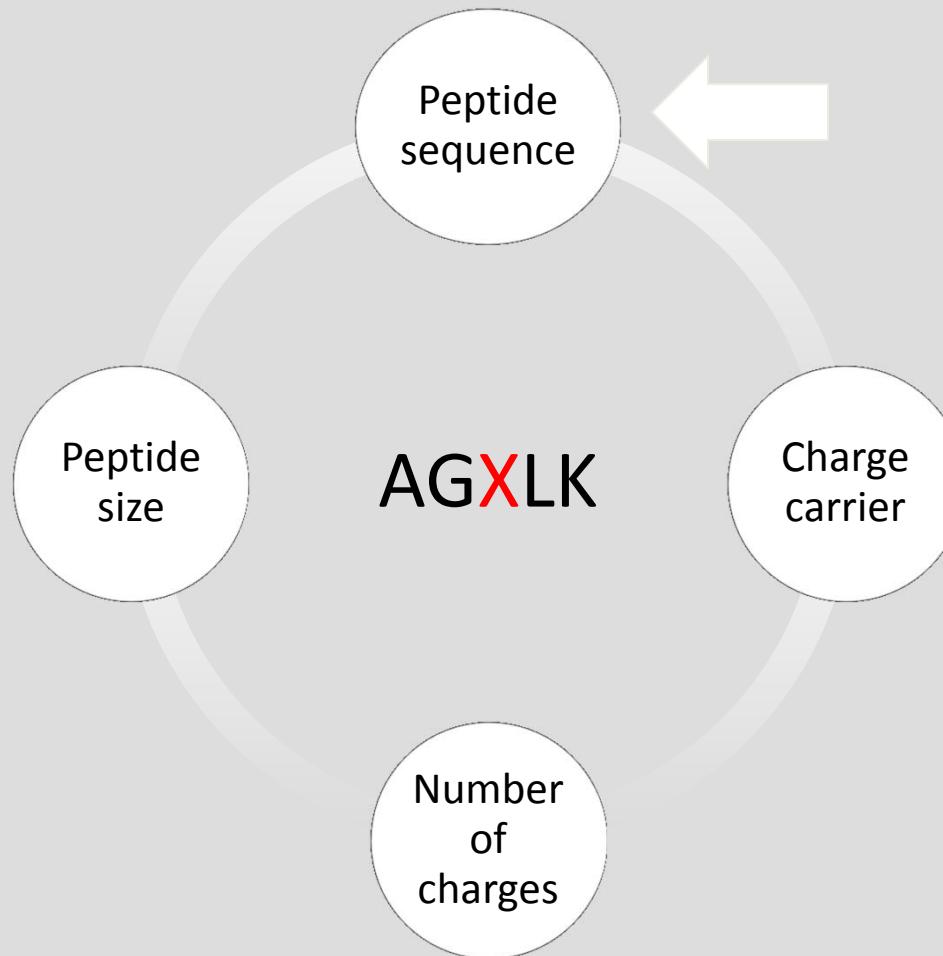
Analysis of the SwedECD database

- Average size of z , w and b fragments as a function of parent peptide size.

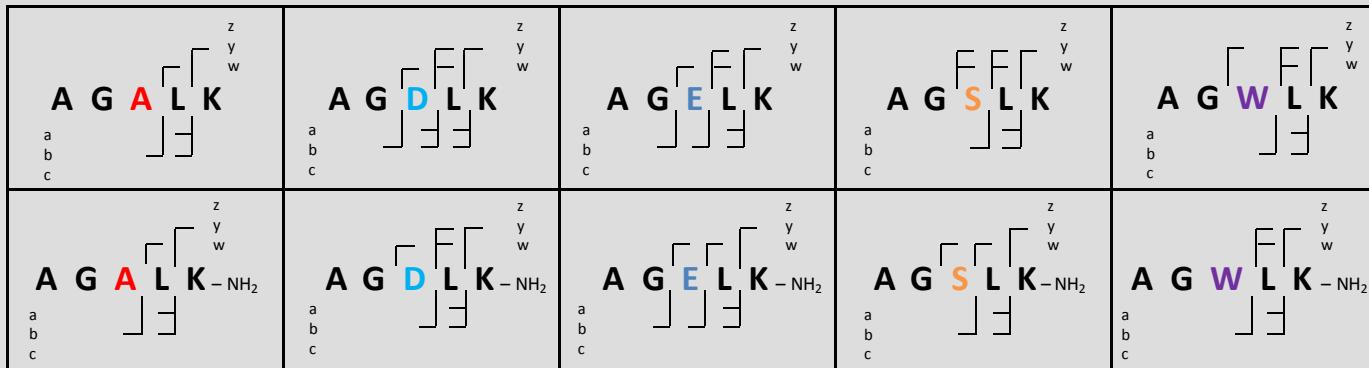


- *Independance of the c/z, b/y and w pathways.*

The AGXLK peptide series



The (AGXLK)H₂²⁺ series

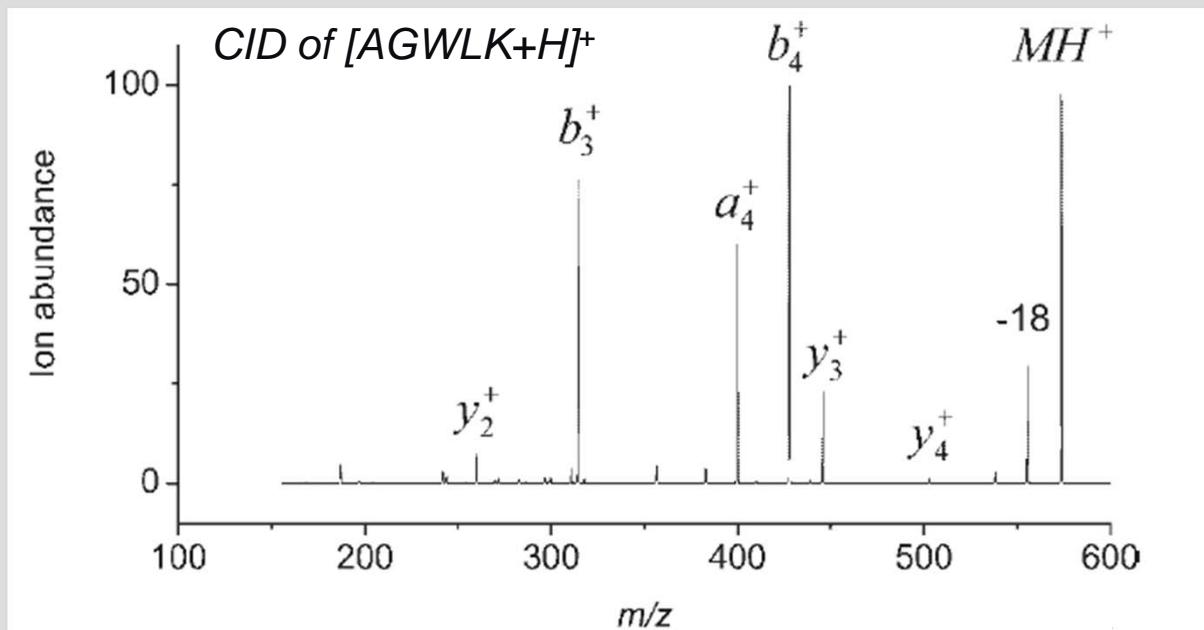


Peptides	b/y	c/z	w	Peptides	b/y	c/z	w
AGALK	73	0	27	AGALK-NH ₂	65	0	35
AGDLK	59	0	41	AGDLK-NH ₂	35	0	65
AGELK	55	0	45	AGELK-NH ₂	48	0	52
AGSLK	53	0	47	AGSLK-NH ₂	64	0	36
AGWLK	76	0	24	AGWLK-NH ₂	68	0	32

- Large b/y and w ions.
- The nature of X does not open the c/z pathway.
- w ions are increased when the side-chain of X can lead to stable radicals.

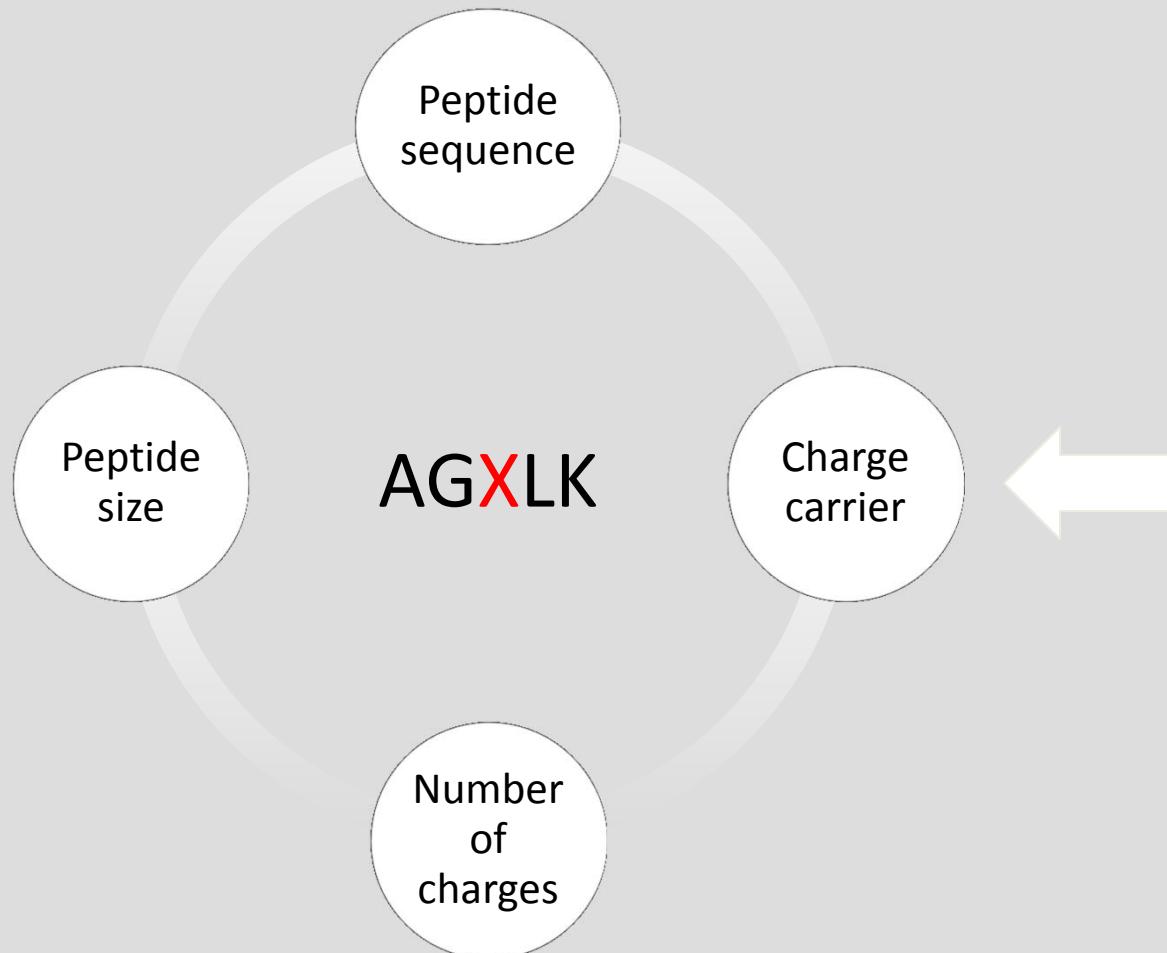
High energy CID spectra

- High and low energy CID of $[M+H]^+$ leads to fragments similar to ECD, except for the w ions.



➤ Could b/y ions be produced from the fragmentation of $[M+H]^+$ precursors?

The AGXLK peptide series

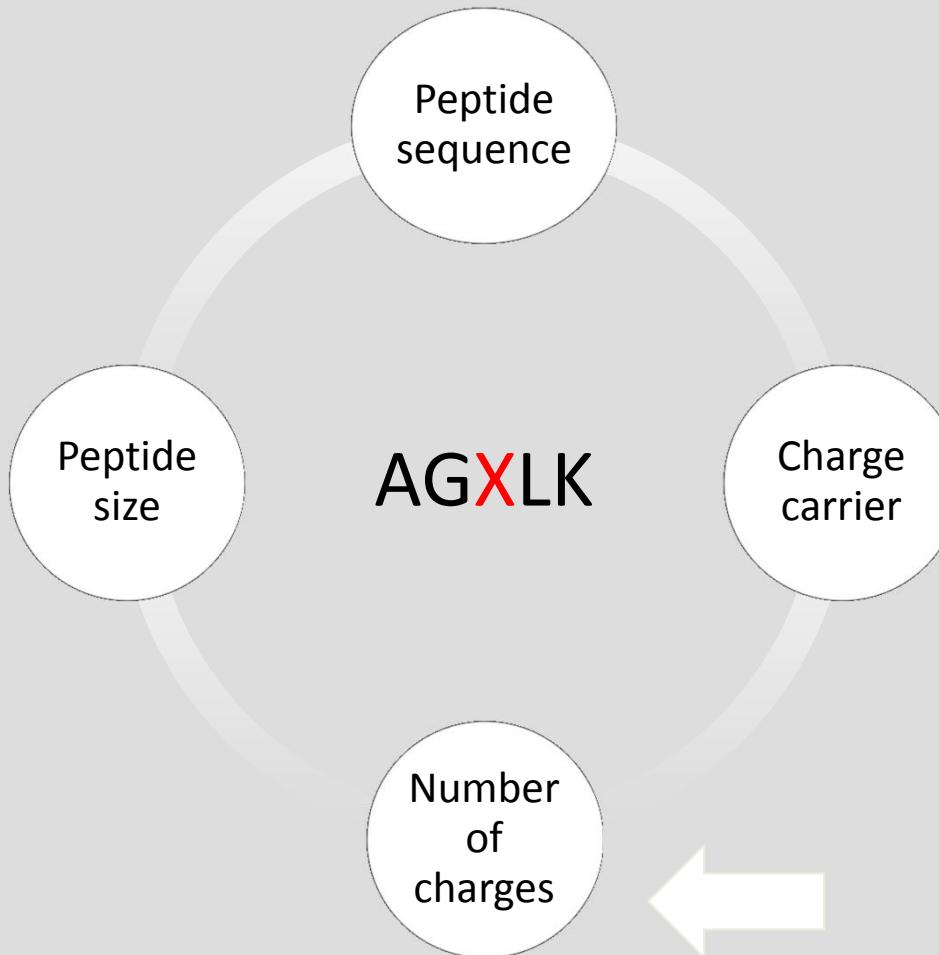


Nature and position of the charged sites

Lys	b/y	c/z	w	Arg	b/y	c/z	w
AGE L K	52	0	48	AGEL R	41	10	49
AGE K L	72	18	10	AGERL	60	14	26
AG K EL	100	0	0	AG R EL	67	33	0
A K GEL	100	0	0	A R GEL	61	39	0
K AGEL	88	12	0	RAGEL	41	59	0

- Lys and Arg containing peptides behave differently
Only Arg leads to significant c/z ions.
 - Major differences which could account for this behavior:
 - Recombination energies (GK: 3.3 eV, GR 2.98 eV)
 - Hydrogen bonding network and charge solvation
- *What factor between energetics and structure is most important?*

The AGXLK peptide series



Number of charged groups

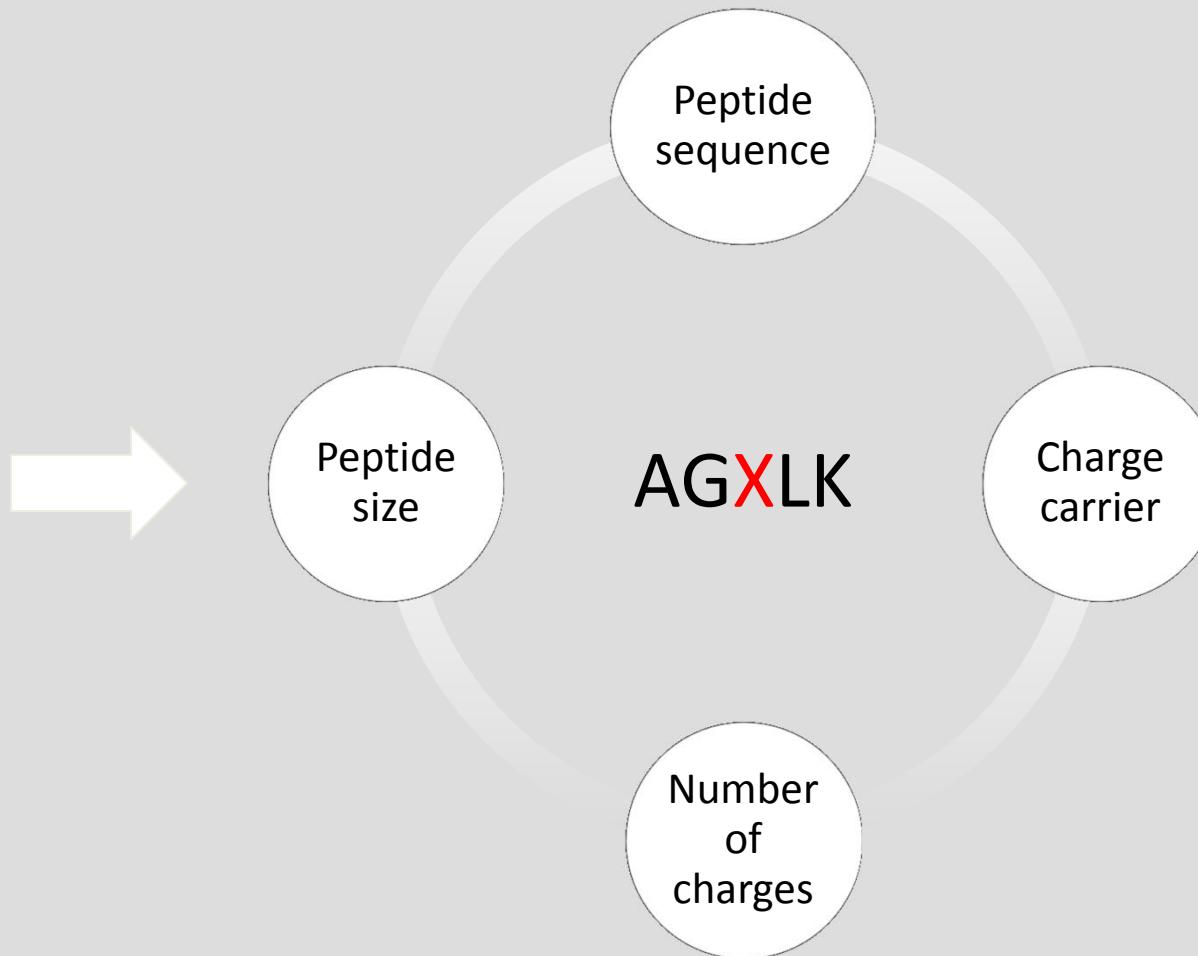
- Comparing 2+ vs 3+ for XGELX peptides (X= Lys or Arg)

Peptides(2+)	b/y	c/z	w	Peptides(3+)	b/y	c/z	w
KGELK	16	42	42	KGELK	72	0	28
KGELR	12	19	69	KGELR	64	0	36
RGELK	15	85	0	RGELK	76	3	21
RGELR	13	74	13	RGELR	68	2	30

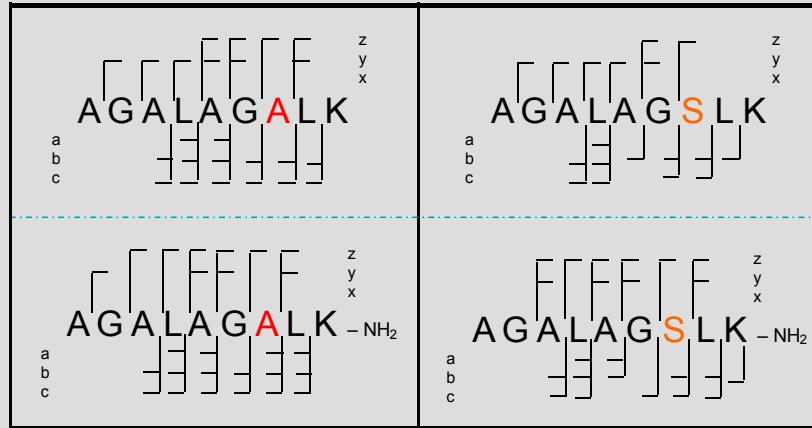
- c/z observed for (2+) but not for (3+)
 - Coulomb repulsion stronger for 2+ than 3+
 - Change in recombination energies
 - Change in structure

- Peut-on tenter de séparer ces deux contributions?
- Que se passe-t-il pour des peptides plus grands, qui auront une structure plus compacte à charge égale?

The AGXLK peptide series

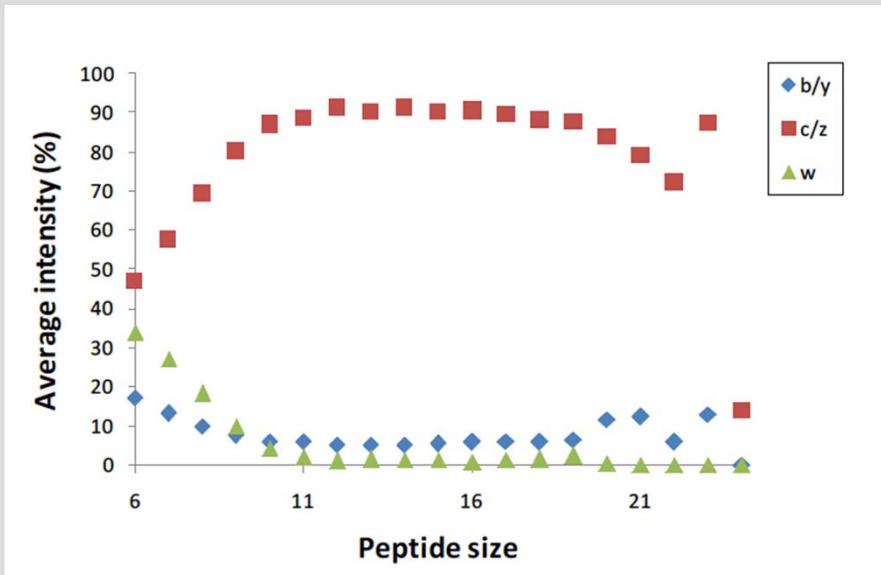


Peptide size



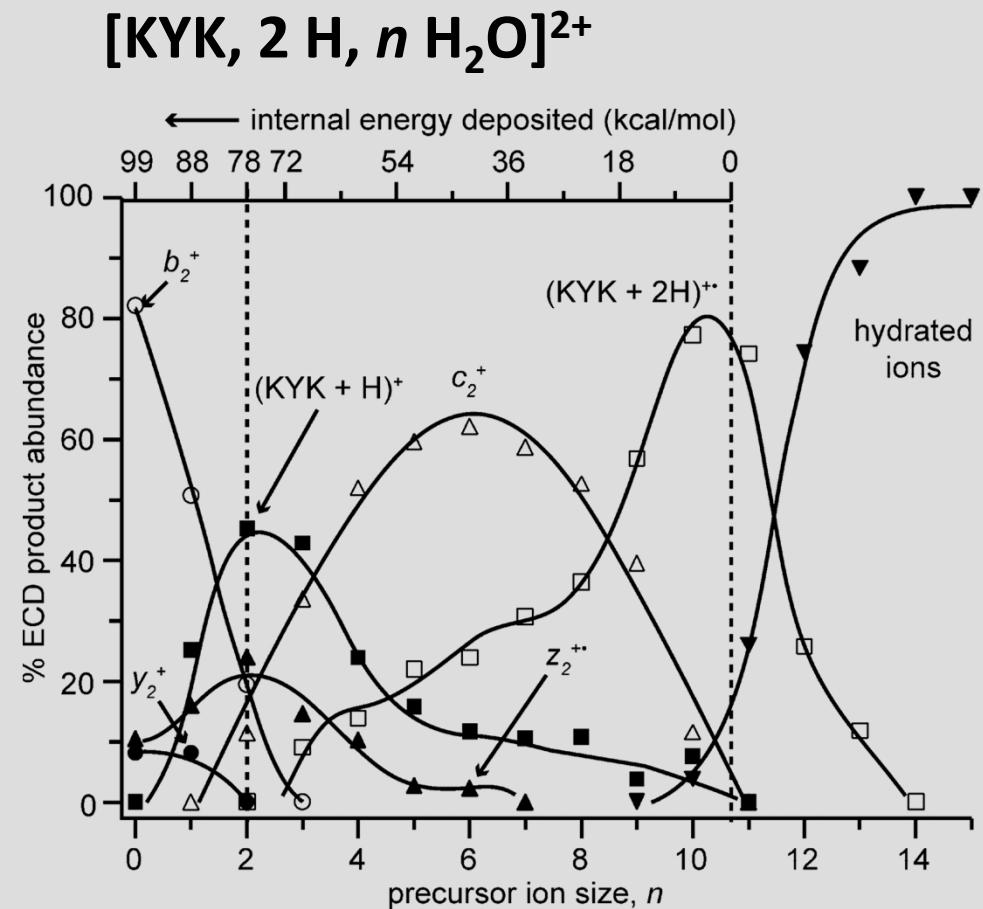
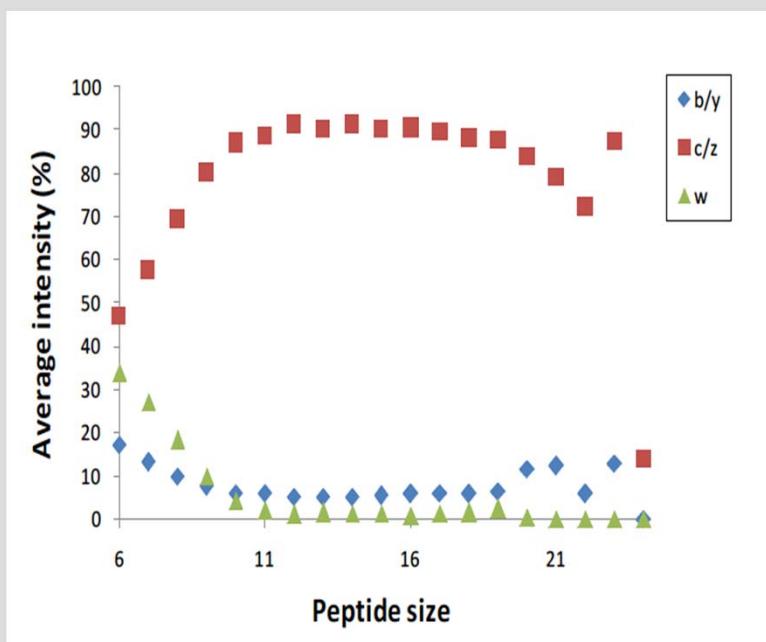
Peptides	b/y	c/z	n'	Peptides	b/y	c/z	n'
AGALAGALK	69	23	8	AGALAGALK-NH ₂	86	11	3
AGALAGSLK	71	14	15	AGALAGSLK-NH ₂	64	25	11

Back to the SwedECD database



- Branching ratios strongly depend on peptide size:
 - For small peptides: dominant *b/y* and *w* fragments
 - For large peptides: dominant *c/z* fragments.

Comparison with nanocalorimetry experiments

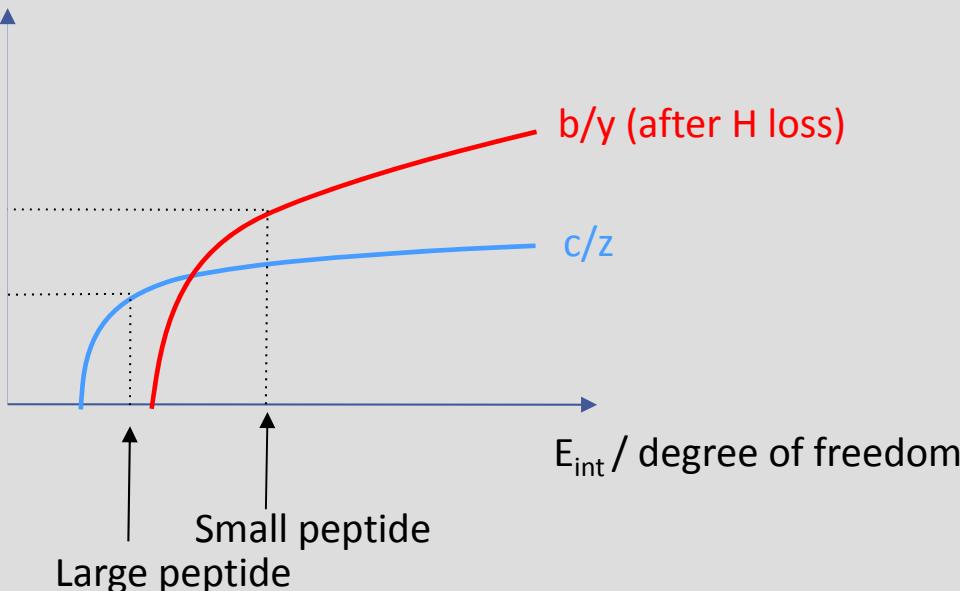


J.S. Prell; et al. J. Am. Chem. Soc. 12680-1268 (2008)9

Energy based model

- Branching ratios depend on the energy density (energy per degree of freedom)

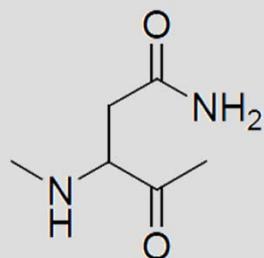
$\log k$



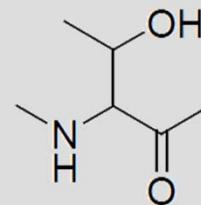
- Can we build a model peptide to test this hypothesis?

Model peptides based on Asn vs Thr

- Asparagine (Asn) will present a carbonyl group that can strongly interact with the charge
 - Should favor the c/z pathway in the structural model.
- Comparison between peptides with the same number of Asn and Thr groups



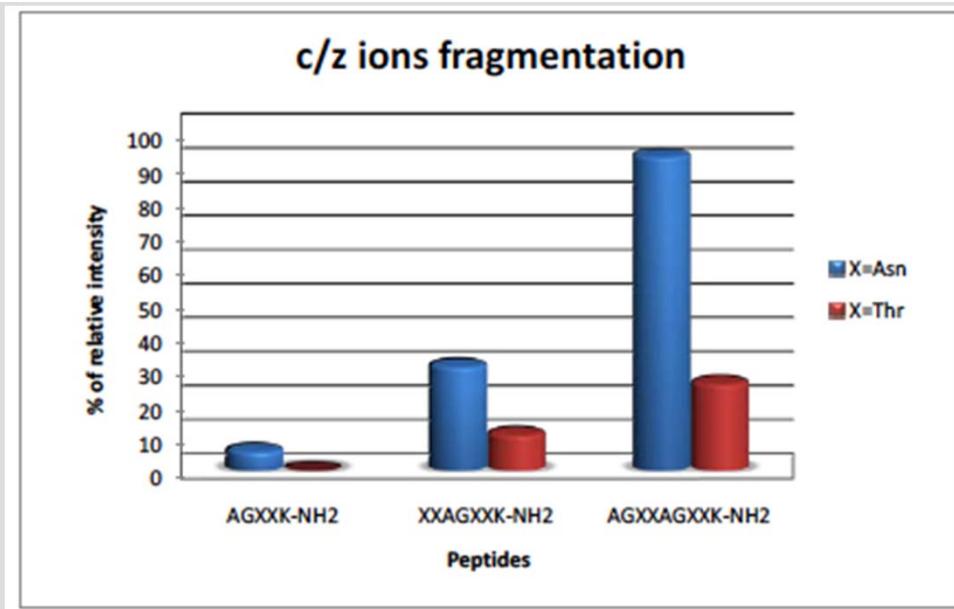
Asparagine (N)



Threonine (T)

Asn/Thr comparison

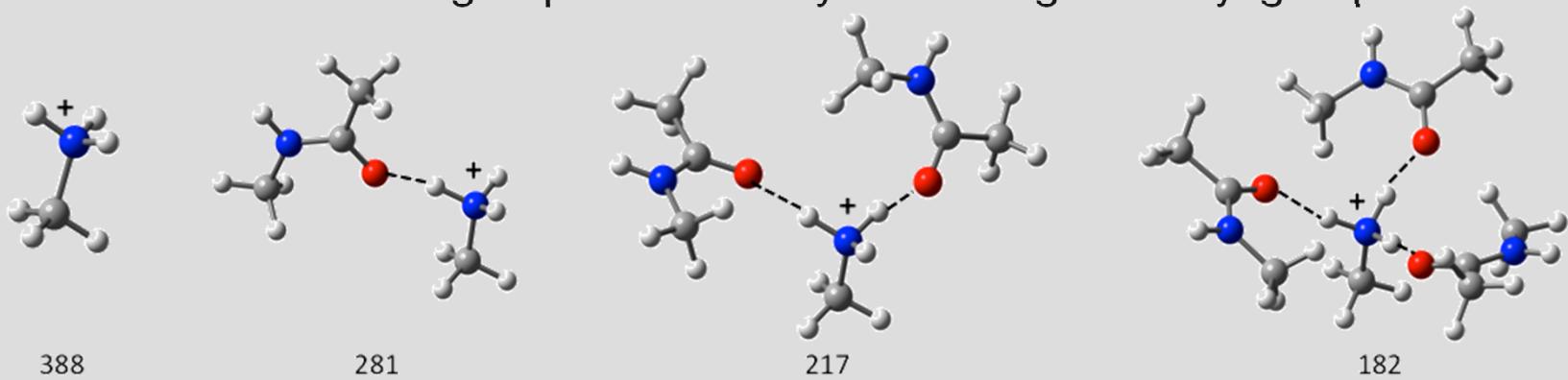
Peptides	b/y	c/z	w	Peptides	b/y	c/z	w
AGNNK-NH ₂	78	6	16	AGTTK-NH ₂	88	0	12
NNAGNNK-NH ₂	42	31	27	TTAGTTK-NH ₂	70	11	19
AGNNAGNNK-NH ₂	2	93	5	AGTTAGTTK-NH ₂	74	26	0



- *c/z pathway increases with the number of asparagine groups.*

Effect of solvation on recombination energies

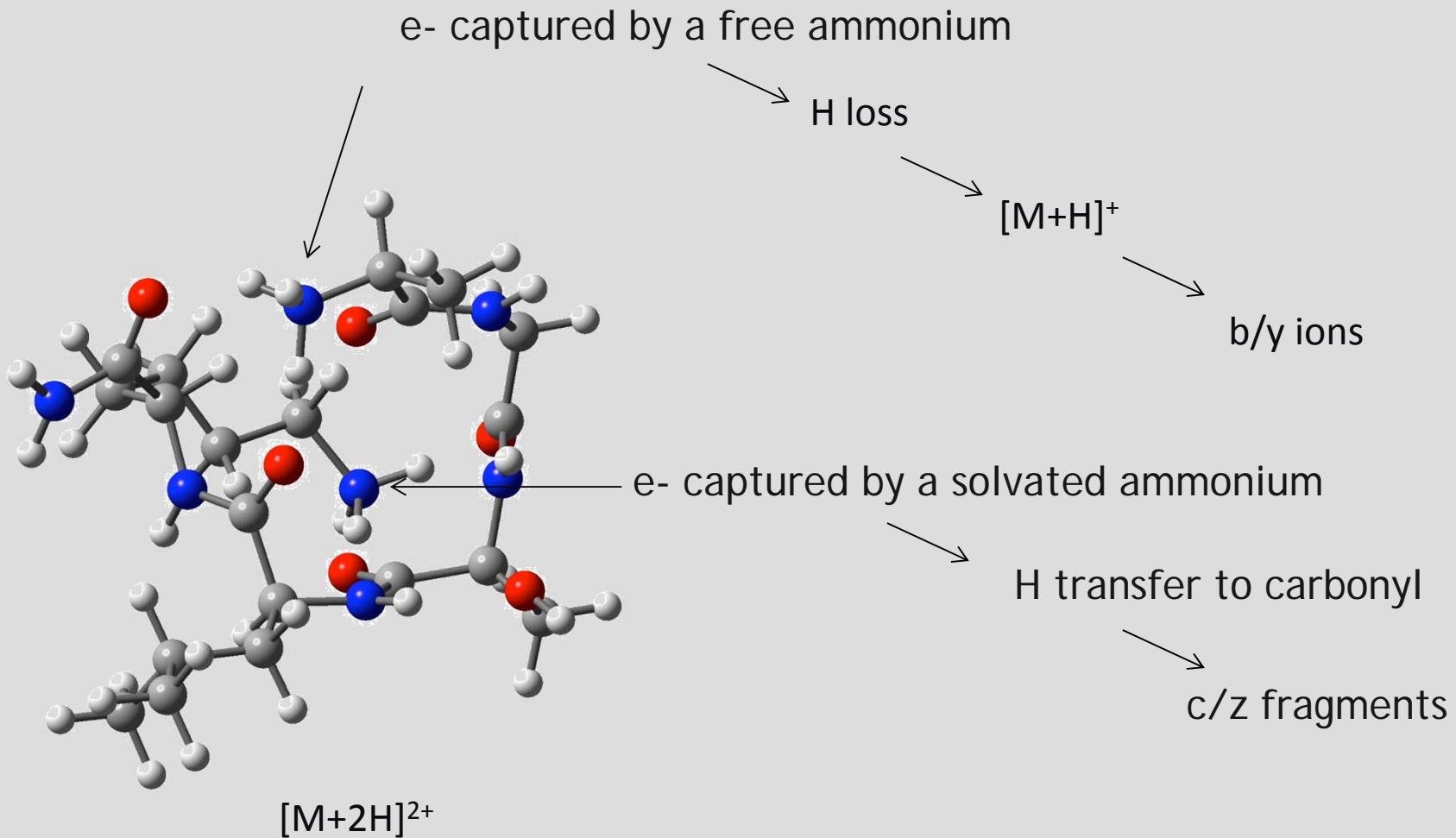
- RE of ammonium groups solvated by increasing carbonyl groups



Computed adiabatic recombination energies in kJ/mol. LC-BLYP/6-311++G(2d,p)

- H bonding can reduced RE up to 2 eV
 - Secondary structure and energetics are linked through RE !

Our current model for peptide fragmentation



Ont participé à ce travail :



J. Chamot - Rooke

G. Frison

R. Hui

C. Malosse



J. Lemaire
P. Maître
J-M. Ortega
B. Rieul
V. Steinmetz
J. Vieira

